



Chapter Ten: Properties of Matter

- **10.1 Density**
- **10.2 Properties of Solids**
- **10.3 Properties of Fluids**
- **10.4 Buoyancy**



Chapter 10.3 Learning Goals

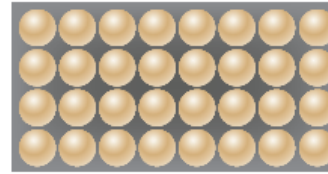
- Explain how pressure is created in fluids.
- Discuss differences between the density of solids and fluids.
- Apply Bernoulli's principle to explain how energy is conserved in fluids.



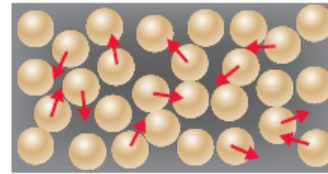
10.3 Properties of Fluids

- A **fluid** is defined as any matter that flows when force is applied.
- Liquids like water or silver are kinds of fluid.

Solid silver

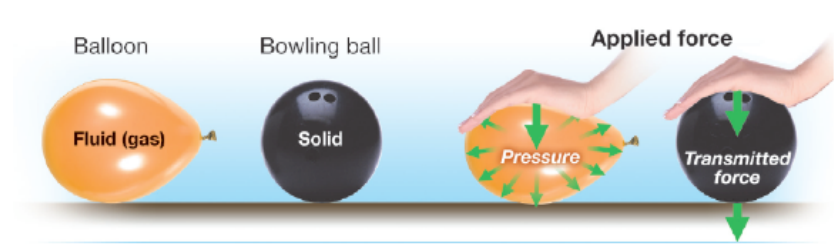


Liquid silver



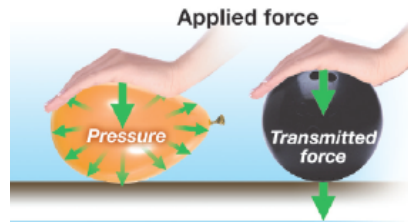
10.3 Forces in fluids

- Forces in fluids are more complicated than forces in solids because fluids can change shape.



10.3 Pressure

- A force applied to a fluid creates pressure.
- Pressure acts in all directions, not just the direction of the applied force.

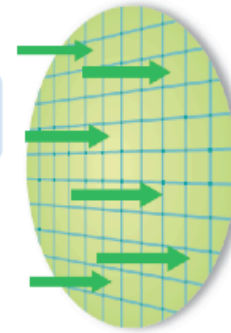




10.3 Units of pressure

- The units of pressure are force divided by area.
- One **psi** is one pound per square inch.

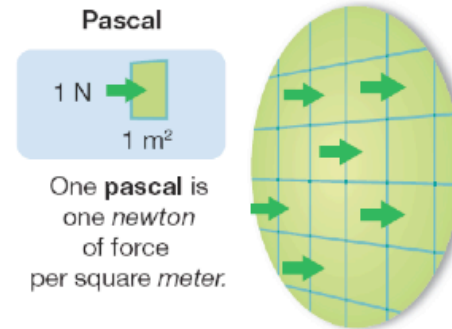
PSi
1 lb → 1 in²
One **PSi** is one *pound* of force per square *inch*.





10.3 Units of pressure

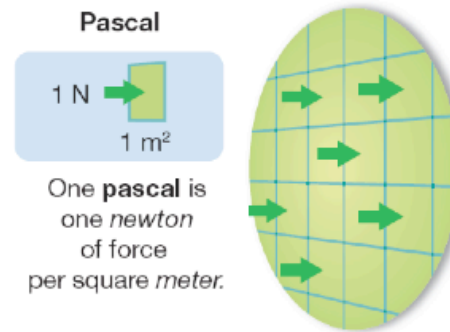
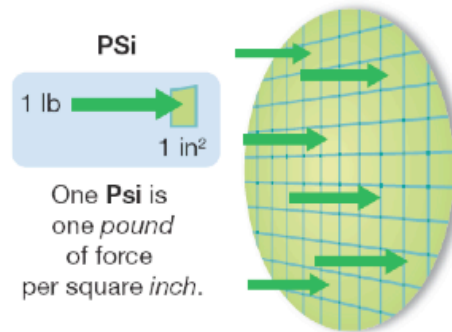
- The S.I. unit of force is the **pascal**.
- One pascal (unit of force) is one newton of force per square meter of area (N/m^2).





Units of Pressure

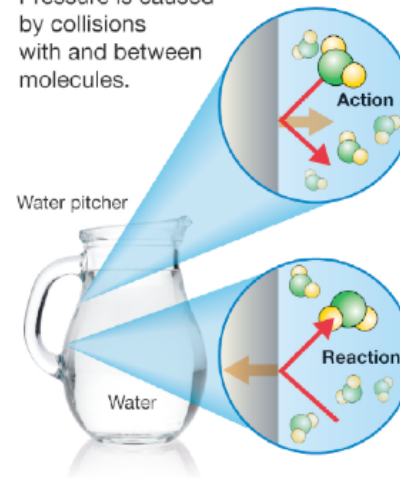
Pressure is force per unit of area.



10.3 Pressure

- On the microscopic level, pressure comes from collisions between atoms.
- Every surface can experience a force from the constant impact of trillions of atoms.
- This force is what we measure as pressure.

Pressure is caused by collisions with and between molecules.





10.3 Pressure



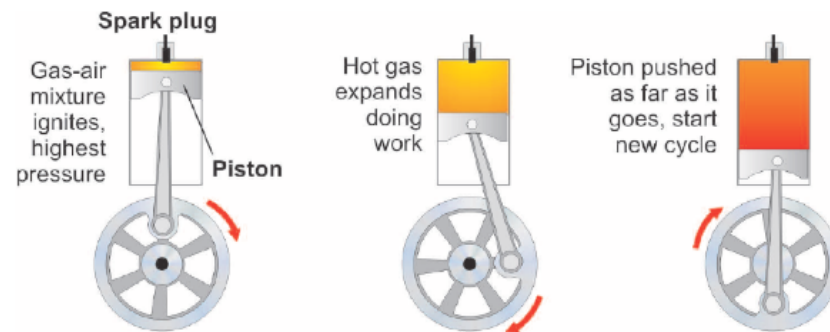
35 pounds acts
on every square inch

- If your car tires are inflated to 35 pounds per square inch (35 psi), then a force of 35 pounds acts on every square inch of area inside the tire.

What might happen if you over-inflate a tire?

10.3 Pressure

- In a car engine high pressure is created by an exploding gasoline-air mixture.



An engine uses pressure in an expanding gas to do work.



10.3 Energy conservation and Bernoulli's Principle



- **Streamlines are imaginary lines drawn to show the flow of fluid.**
- **Bernoulli's principle tells us that the energy of any sample of fluid moving along a streamline is constant.**



10.3 Bernoulli's Principle

Form of energy	Potential energy	Kinetic energy	Pressure energy	=	Constant along any streamline in a fluid
Variable	height	speed	pressure		

- **Bernoulli's principle says the three variables of height, speed and pressure are related by energy conservation.**



10.3 Three Variables and Bernoulli's Principle

Form of energy	Potential energy	Kinetic energy	Pressure energy	=	Constant along any streamline in a fluid
Variable	height	speed	pressure		

- **If one variable increases along a streamline, at least one of the other two must decrease.**
- **For example, if speed goes up, pressure goes down.**



10.3 Bernoulli's Principle

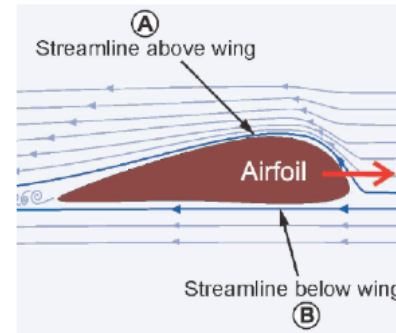
Form of energy	Potential energy	Kinetic energy	Pressure energy	=	Constant along any streamline in a fluid
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- **Bernoulli's principle says the three variables of height, speed and pressure are related by energy conservation.**



10.3 The air foil

- One of the most important applications of Bernoulli's principle is the airfoil shape of wings on a plane.
- When a plane is moving, the pressure on the top surface of the wings is lower than the pressure beneath the wings.
- The difference in pressure is what creates the lift force that supports the plane in the air.





10.3 Boyle's Law

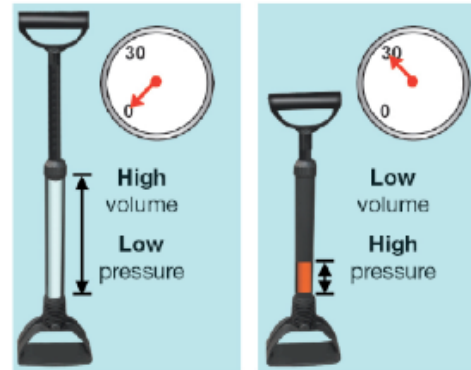


- When you squeeze a fixed quantity of gas into a smaller volume the pressure goes up.
- This rule is known as Boyle's law.



10.3 Boyle's Law

- The formula for Boyle's law relates the pressure and volume of gas.
- If the mass and temperature are kept constant, the product of the pressure multiplied by the volume stays the same.





Solving Problems: Boyle's Law

BOYLE'S LAW

$$\begin{array}{ccccccc} & & \text{Initial volume} & & \text{New pressure} & & \\ & & | & & | & & \\ \text{Initial pressure} & - & P_1 & V_1 & = & P_2 & V_2 & - \text{New volume} \\ & & | & & & & & \\ & & \text{Initial volume} & & \text{New volume} & & & \end{array}$$

Mass and temperature remain constant



Solving Problems

A kit used to fix flat tires consists of an aerosol can containing compressed air and a patch to seal the hole in the tire.

Suppose 5 liters of air at atmospheric pressure (1 atm) is compressed into a 0.5 liter aerosol can. What is the pressure of the compressed air in the can?

Assume no change in temperature or mass.



Solving Problems

1. Looking for:

- ...final pressure in atmospheres (P_2)

2. Given

- ... $V_1 = 5 \text{ L}$, $P_1 = 1 \text{ atm}$, $V_2 = .5 \text{ L}$

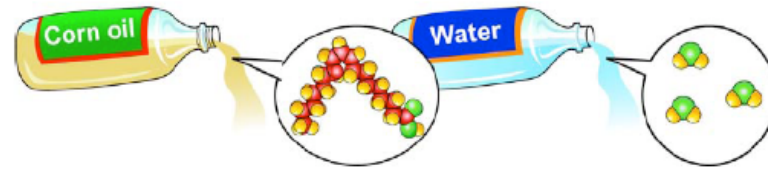
3. Relationships:

- Boyle's Law: $P_1V_1 = P_2V_2$

4. Solution

- Rearrange equation so $P_2 = P_1V_1 / V_2$
- $P_2 = 1\text{atm} \times 5.0 \text{ L} / 0.5 \text{ L} = \mathbf{10 \text{ atm}}$.

10.3 Viscosity



- **Viscosity** is the property of fluids that causes friction.
- Viscosity is determined in large part by the shape and size of the particles in a liquid.



10.3 Viscosity and temperature



- As the temperature of a liquid increases, the viscosity of a liquid decreases.
- Increasing the kinetic energy of the substance allows the particles to slide past one another more easily.

Investigation 10B

Buoyancy

- **Key Question:**
Can you make a clay boat?

