# CHAPTER 16 What is Matter?







# SECTION 16.1 CLASSIFYING MATTER

 Matter is a term that describes anything that has mass and volume (takes up space).

# Matter can be divided into 2 categories:

1. Mixtures

2. Substances

## 1. MIXTURES

 Mixtures contain more than one kind of matter.

There are 2 kinds of mixtures.

**INGREDIENTS** carbonated water corn syrup caramel color phosphoric acid natural flavors caffeine



## HOMOGENEOUS MIXTURE

• A mixture that is the same throughout.

• Examples:

coffee

brass

air







#### HETEROGENEOUS MIXTURE

- A mixture in which different samples are <u>not</u> necessarily the same.
- Examples:chicken soupsalad dressing

# All mixtures share one common property:

• They can be separated by ordinary physical methods.

Examples: sorting, filtering, solubility, magnetic properties, density, heating, or cooling.

## 2. SUBSTANCES

- Substances cannot be separated by physical methods.
- •Examples:

salt



gold



#### **ELEMENTS**

- Some substances are elements and contain only one kind of matter (atoms). They cannot be separated by ordinary chemical or physical methods.
- Examples: silver



helium



See the periodic table for a complete list of the elements!

## COMPOUNDS

- Compounds contain two or more kinds of matter (atoms), but cannot be separated without changing the substance.
- Examples:

water

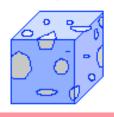




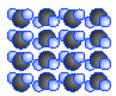
So... all mixtures are either homogeneous or heterogeneous and all substances are either elements or compounds.

#### **Types of Matter**









Heterogeneous mixture

Element

Compound

Type of matter	Definition	Examples
Homogeneous mixture	A mixture that contains more than one type of matter and is the same throughout.	soda pop, air, chocolate ice cream
Heterogeneous mixture	A mixture that contains more than one type of matter and is not the same throughout.	chicken soup, soil, fudge ripple ice cream
Element	A substance that contains only one type of atom.	copper metal, oxygen gas, liquid nitrogen
Compound	A substance that contains more than one type of atom.	table salt, rust (iron oxide), carbon dioxide gas

#### **MATTER MIXTURES SUBSTANCES HOMOGENEOUS ELEMENTS COMPOUNDS HETEROGENEOUS MIXTURES** MIXTURES **@@@@** Heterogeneous Homogeneous Compound Element mixture mixture